CRYSTAL CHEMISTRY AND SUPERCONDUCTIVITY OF PRESSURE-INDUCED PHASES IN THE In-Te SYSTEM

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Abstract—In the In—Te system, a pressure-induced NaCl-type phase exists in the region In_{0.80}Te to In_{1.15}Te. Superconductivity exists in the whole range, with the maximum transition temperature occurring for the stoichiometric InTe. A hypothesis previously given for the metallic behavior of this phase and the decrease of transition temperature on either side of the stoichiometric InTe is further elaborated. It is proposed that the transformation to the NaCl-type structure removes the structural constraint on electron transfer existing in the normal InTe. A method for calculating carrier concentration is given and it is shown that the superconducting transition temperature is a function of carrier concentration.

At least two other pressure-induced In–Te phases exist. One is In₃Te₄ which becomes superconducting at 1·25–1·15°K. In₃Te₄ has the anti-Sn₄As₃ structure with seven atoms in a rhombohedral unit cell, all lying on the threefold axis. The positional parameters of the atoms and interatomic distances are given and the coordination shown to be related to that found in the NaCl-type structure. Pressure-temperature experiments on normal In₂Te₃ indicate that a pressure-induced phase with this (or approximately this) composition cannot be obtained metastably without the presence of the In₃Te₄ and an unidentified phase. The pressure-induced In₂Te₃ phase has the well-known Bi₂Te₃ structure and is closely related to the In₃Te₄ structure. Superconductivity tests and X-ray diffraction investigation lead to the conclusion that the In₃Te₄ phases occurring with the In₂Te₃ phase are sometimes not stoichiometric and in such cases usually contain excess tellurium.

INTRODUCTION

In a recent letter, (1) we reported the occurrence and superconducting behavior of pressure-induced vacancy structures of the NaCl-type in the In—Te system. It was proposed that the metallic behavior of the NaCl-type InTe resulted from the removal of the structurally imposed constraint in the normal phase, on electron transfer from In+to In³⁺ ions. This proposal has led to successful prediction (2) of other superconductors with NaCl-type and a related structure and of existence of solid solution regions in NaCl-type phases of binary systems.

The behavior of the In-Te system at high pressures is quite complex and we shall discuss in this paper only those aspects of it which are fairly clear.

EXPERIMENTAL

Appropriate amounts of the constituent elements in each case were melted together in an evacuated sealed

fused silica tube. The resulting product was ground and mixed thoroughly to ensure homogeneity. X-ray powder photographs always showed the mixture of normal compounds expected from the phase diagram reported by KLEMM and VOGEL.⁽³⁾ Specimens were packed into Ta containers and subjected to pressure and heating as described elsewhere.⁽⁴⁾ The NaCl-type compounds were obtained usually by using 30 kbar and 400–500°C for various lengths of time not exceeding 4 hr. Pressures and temperatures for other phases will be given in later sections.

X-ray powder photographs (CuK radiation) were taken of all specimens subjected to high pressure and in some cases diffractometer patterns were obtained. Single crystal X-ray data were also obtained in some cases to be discussed later.

Superconductivity tests were made by the method of Schawlow and Devlin. (5)

THE NaCl TYPE PHASES

Results

Since our earlier publication⁽¹⁾ on the Te-rich NaCl-type compounds, we have obtained In-rich pressure-induced NaCl-type phases to approximately In_{1·15}Te. It has also been possible to extend

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